ORGANIC 1

1.

(a) Only single bonds between carbon atoms 1 [1]
(b) (i) Water
Carbon dioxide [2]
(ii) No carbon or soot produced
No nitrogen oxides produced
No sulphur oxides produced
No carbon monoxide produced
[2]

2.

any two of: [2]

family of similar (organic) compounds /

with similar chemicalproperties /

presence of same functional group /

same general formula /

allow:compounds with a trend in physical properties

allow:difference of CH2between one member and another

3.				
		One general formula / same general formula; differ by CH ₂ ;		
		similar chemical properties; gradual change in physical properties; Award [1] for any two from last three		
		functional group: atom or group of atoms responsible for the characteristic reactions of the molecule / homologous series;		3
4.				
	(a)	(i) boiling point increases as number of carbons increases / OWTTE increased surface area / greater Van der Waals' forces / increase		<i>M</i> r /
			increased intermolecular forces / OWTTE;	2
		(ii)	exothermic / energy released / products have less energy than reactants;	1
	(b)	carbon dioxide; water;		2
		Accept formulas.		

Total [5]

[2]

[5]

(a) (i) breaking down of molecules substances using heat [1] (ii) substance which speeds up a reaction [1] NOT: alters/changes rate of reaction NOT: speeds up and slows down rate (b) ethene/ethylene [1] NOT: formula (c) (i) paraffin [1] (ii) 4000g/4kg [1] (correct unit needed) (iii) C₂H₄; H₂ [2] (d) (i) two units polymerised with continuation bonds at either end and hydrogen atoms drawn [1] ALLOW: _CH₂CH₂CH₂CH₂ _ ALLOW: $[_CH_2CH_2 _] _ n$ ALLOW: [_CH₂] _ n (ii) addition (polymerisation) [1]

[Total 10m]