<u>233/3</u> <u>Chemistry Practical</u>

SCHOOL BASED FORM 4 EXAMINATIONS JULY – AUGUST 2017

Confidential

<u>In additiion to the fittings and apparatus found ina a Chemistry Laboratory each</u> <u>student will require the following.</u>

- 1. 1.5g of solid T
- 2. 250 cm³ of solution B
- 3. 150 cm^3 of solution R
- 4. 3 conical flasks
- 5. 100cm³ measuring cylinder
- 6. Distilled water in a wash bottle
- 7. Burette
- 8. Pipette
- 9. Filter funnel
- 10. Filter paper
- 11. Spatula
- 12. 25cm³ measuring cylinder
- 13. Solid S
- 14. 5 test tubes in a rack
- 15. 1 boiling tube
- 16. Test tube holder
- 17. Universal indicator paper
- 18. pH chart
- 19. Liquid A
- 20. Ethanol
- 21. Red and blue litmus paper

Access to

- 1. Phenolphthalein indicator
- 2. Means of heating
- 3. Universal indicator solution
- 4. Ammonia solution
- 5. Potassium Iodide solution
- 6. Barium nitrate solution
- 7. Aqueous nitric (V) acid
- 8. Acidified potassium manganate (VII) solution
- 9. Bromine water
- 10. Acidified potassium dichromate (VI) solution

NOTES

- 1. Solid T is Benzoic acid
- 2. Solution B is 0.02M NaOH
- 3. Solution R is $0.01MH_2SO_4$
- 4. Solid S is hydrated AluminiumsulphateAL₂(SO₄)₃. 6H₂O.
- 5. Liquid A is distilled water.